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answer. 157.5 g 106.5 g 51 g Yes. Mass of reactants equals mass of products. Section 3.2 Assessment page 79 10. Classify each example as a physical change or a chemical change. a. crushing an aluminum can physical b. recycling used aluminum cans to make new aluminum cans physical

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c. aluminum combining with oxygen to form aluminum oxide chemical 11.

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Section 10.1 Assessment page 324 7. Explain why chemists use the mole. Chemists use the mole because it is a convenient way of knowing how many representative particles are in a sample. 8. State the mathematical relationship between Avogadro's number and 1 mol. One mole contains 6.02  $\times 10^{23}$  representative particles. 9.

### **The Mole**

Thinking Critically Answer the following questions. Chemistry: Matter and Change • Chapter 5 Chapter Assessment 1. A radio station has a frequency of 103.7 MHz. (1 MHz  $10^6$  s<sup>-1</sup>) What is the wavelength of the radiation emitted by the station? Indicate where this wavelength falls on the electromagnetic spectrum shown below.  $c = 3.00 \times 10^8$  m/s ...

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