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Many of water's unusual properties occur because of the attraction among its polar molecules. The properties of water include cohesion, adhesion, capillary action, surface tension, the ability to dissolve many substances, and high specific heat. The tendency for water molecules to form weak bonds and stick to each other is called cohesion.

The Properties of Water - Marine Science- Mrs. Ogo

Properties of water post lab quesitons Name _____ Class: _____ 1. Write the chemical equation for water, and draw a Stick model of a water molecule 2. Sketch a space filling model of a water molecule, show polarity by label positive and negatively charged regions of the molecule. 3. What is a polar molecule? 4.

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As water freezes, air becomes trapped between the hydrogen bonds of water molecules. As water freezes, it takes up more oxygen from the atmosphere, causing it to have a greater buoyancy.

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