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Round off each measurement to three significant figures. A) 98.473 L B) 0.000 763 21 cg C) 57.048 m D) 12.17°C

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Chapter 3 - Scientific Measurement. Chapter 3 - Scientific Measurement. Jennie L. Borders. Section 3.1 - Measurements and Their Uncertainty. A measurement is a quantity that has both a number and a unit. The unit typically used in the sciences are those of the International System of Measurements (SI). In scientific notation, a given number is written as the product of two numbers: a coefficient and 10 raised to a power.

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11th - 12th grade. 411 times. Chemistry. 74% average accuracy. a year ago. cshell76. 1. Save. Edit. ... Answer not given. Tags: Question 9 . SURVEY . 30 seconds . Q. All the digits that can be known precisely in a measurement, plus a last estimated digit is called.

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Pete's three measurements are 20.9 cm, 21.0 cm, and 21.0 cm. Calculate the average value of his measurements and express the answer with the correct number of significant figures. 3. Multiply the answer to problem 1 by the answer to problem 2.

Chapter 3 Practice Problems Key | Significant Figures ...

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Scientific Measurement Quantifying Matter 3.1 using and expressing Measurements essential Understanding In science, measurements must be accurate, precise, and written to the correct number of significant figures. reading Strategy Venn Diagram A Venn diagram is a useful tool in visually organizing related information.

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thoughts to the spot is just not simple.

Chapter 3 Scientific Measurement Worksheet Answer Key

...

PEP - Chemistry/ Chapter 3 Answer Key 4 15. Solve the following and express each answer in scientific notation. = 6.6×10^a .
 $(5.3 \times 10^4) + (1.3 \times 10^4) = 4.0 \times 10^b$.
 $(7.2 \times 10^{-4}) / (1.8 \times 10^3) = 8.7 \times 10^c$.
 $10^4 \times 10^{-3} \times 10^6 = 10^7$.
 $d. (9.12 \times 10^{-1}) - (4.7 \times 10^{-2}) = 10^e$.
 $(5.4 \times 10^4) \times (3.5 \times 10^9) = 1.9 \times 10^{14}$.
f.

Chapter 3 Scientific Measurement

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SCIENTIFIC MEASUREMENT Class Column B density precision conversion factor Chapter Test B A. Matching Match each term in Column B with the correct description in Column A Write the letter of the correct term on the line. 2. 3. 8. 10. Column A how close a single measurement comes to the actual value of whatever is being measured

Cardinal Newman High School

Chapter 3 " Scientific Measurement " 2. ... Answer 22.68 m 2 23 m 2 100.0 g ÷ 23.7 cm 3 4.219409283 g/cm 3 4.22 g/cm 3 0.02 cm x 2.371 cm 0.04742 cm 2 0.05 cm 2 710 m ÷ 3.0 s 236.6666667 m/s 240 m/s 1818.2 lb x 3.23 ft 5872.786 lb·ft 5870 lb·ft 1.030 g x 2.87 mL 2.9561 g/mL 2.96 g/mL

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