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meaning they have a numerical value. A measurement is a quantity that has both a number and a unit. The unit typically used in the sciences are those of the International System of Measurements (SI).

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Chapter 3: Scientific Measurement. STUDY. Flashcards. Learn. Write. Spell. Test. PLAY. Match. Gravity. Created by. kbenn06. pages 63-94. Terms in this set (21) How many significant figures are in each measurement? A) 60 s B) 47.70 g of copper C) 25 computers. A) 1 B) 4 C) 2. Round off each measurement to three significant figures.

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1. Every nonzero digit in a reported measurement is assumed to be significant
2. Zeros appearing between nonzero digits are significant
3. Leftmost zeros appearing in front of nonzero digits are not significant. They act like placeholders. By writing the measurements in scientific notation, you eliminate such placeholder zeros
4.

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Chapter 3: Scientific Measurement. Repp Chemistry A. STUDY. PLAY. absolute zero. the zero point on the Kelvin temperature scale, equivalent to -273.15°C . accepted value. a quantity used by general agreement of the scientific community. accuracy. the closeness of a measurement to the true value of what is being measured.

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Chapter 3 Scientific Measurement Chapter 3: Scientific Measurement. Measurement. scientific notation. accuracy. precision. a quantity that has both a number and unit. an expression of numbers in the form $m \times 10^n$ where m is equal....

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a measure of how close a measurement comes to the actual or tr.... chemistry chapter 3 scientific measurement Page 6/24

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Scientific Measurement Quantifying Matter 3.1 using and expressing Measurements essential Understanding In science, measurements must be accurate, precise, and written to the correct number of significant figures. reading Strategy Venn Diagram A Venn diagram is a useful tool in visually organizing related

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Chapter 3 Scientific Measurement Review The pressure reading from a barometer is 742 mm Hg. Express this reading in kilopascals, kPa. (Use dimensional analysis to solve. 760 mm Hg = 1.013 x 10⁵ Pa) 98.9 kPa Express 0.000 000 000 154 m in scientific notation. 1.54 x 10⁻¹⁰ m What is 5928 km expressed in scientific notation?

Chapter 3 Scientific Measurement Review

Chapter 3 - Scientific Measurement Learning Goals: 1. Students will understand how to use scientific measurement as a method of quantifying matter. 2. Students will be able to represent measurements in scientific notation, identify differences in accuracy and precision, and illustrate significance of measurements.

Chemistry I Chapter 3 - Scientific Measurement

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Chapter 3 Scientific Measurement Review. Chapter 3. Scientific Measurement. Review. A rectangular block of lead measures 20.0 mm x 30.0 mm x 45.0mm. If the density of Pb is 11.34 g/cm³, calculate the mass of the block. 306 g As the mass of a substance increases, the density of a given volume of that substance ... increases What is the volume of 344 g of ice, if the density is 0.92 g/mL? 370 mL The pressure reading from a barometer is 742 mm Hg.

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