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Chapter 6  
Chemical Bonding  
Section 2  
Covalent Answer  
Key

**Chapter 6**  
**Chemical**  
**Bonding**  
**Section 2**  
**Covalent**  
**Answer Key**

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Chapter 6

Chemical Bonding

Section 2

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## Chapter 6

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### **Chapter 6 Chemical Bonding Section**

CHAPTER 6 REVIEW

Chemical Bonding

SECTION 1 SHORT

ANSWER Answer the following questions in the space provided. 1.

a A chemical bond between atoms results from the attraction between the valence

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## Chapter 6

Chemical Bonding

Section 2

Covalent Bonding and Molecular  
Lewis structures 2. b A

covalent bond consists  
of (a) a shared  
electron.

### **6 Chemical Bonding - Effingham County School District**

Chemical Bonding

Section 1 Introduction

to Chemical Bonding

Section 2 Covalent

Bonding and Molecular

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## Chapter 6

Chemical Bonding

Section 3  
Compounds section 3  
Ionic Bonding and Ionic

Compounds section 4  
Metallic Bonding

section 5 Molecular  
Geometry CHAPTER 6.

Chemistry

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Introduction to

Chemical Bonding Key

Terms chemical bond

nonpolar-covalent

bond ...

## **CHAPTER 6 hemical Bonding**

*Page 7/26*

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## Chapter 6

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Section 2

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CHAPTER 6 REVIEW  
Chemical Bonding  
SECTION 1 SHORT  
ANSWER Answer the  
following questions in  
the space provided. 1.

a A chemical bond  
between atoms results  
from the attraction  
between the valence  
electrons and of

different atoms. (a)  
nuclei (c) isotopes (b)  
inner electrons (d)

Lewis structures 2. b A

**Chapter 6 Chemical**



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Chemical Bonding

**Bonds -**

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Modern Chemistry 9  
Chemical Bonding

CHAPTER 6 STUDY

GUIDE Chemical

Bonding SECTION 3

IONIC BONDING AND

IONIC COMPOUNDS

SHORT ANSWER

Answer the following  
questions in the space

provided. 1. \_\_\_\_\_ The

notation for sodium

chloride, NaCl, stands

for one (a) formula

unit. (c) crystal.

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Bonding - mchsapch  
emistry.com**

'CHAPTER 6 CHEMICAL  
BONDING Section 1

Introduction to May  
8th, 2018 - CHAPTER 6  
CHEMICAL BONDING

Section 1 Introduction  
to Chemical Bonding A  
chemical bond is a  
mutual electrical  
attraction between the  
nuclei and valence'

'Chapter 6 Review  
Chemical Bonding

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Chapter 6  
Chemical Bonding  
Answers Pdf  
Section 2

**Chapter 6 Review  
Chemical Bonding**

Chapter 6 - Chemical Bonds. Jennie L. Borders. Standards. SPS1. Students will investigate our current understanding of the atom. b. Compare and contrast ionic and covalent bonds in terms of electron movement. SPS2. Students will explore the nature of matter,

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## Chapter 6

Chemical Bonding

its classification and its

system for naming

types of matter.

Section 2  
Covalent Answer

### Chapter 6 - Chemical Bonds

CHAPTER 6 REVIEW

Chemical Bonding

SECTION 4 SHORT

ANSWER Answer the

following questions in

the space provided. 1.

\_\_\_\_\_ In metals, the valence electrons are considered to be (a) attached to particular positive ions, (c)

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Chemical Bonding

Section 2  
Covalent Answer  
Key

immobile. (b) shared by all surrounding atoms. (d) involved in covalent bonds. 2.

Key The fact that metals are malleable and ionic crystals are brittle is best explained in terms of their (a) chemical bonds.

### **CHAPTER 6 REVIEW** **Chemical Bonding**

A hydrogen bond is a dipole - dipole attraction between a partially positive

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## Chapter 6

Chemical Bonding

Section 2

Covalent Answer  
Key

hydrogen atom and the unshared electron pair of a strongly electronegative atom such as O, N, or F.

Unlike ionic or covalent bonds, in which electrons are given up or shared, the hydrogen bond is a weaker attraction.

### **Chapter 6 Review: Chemical Bonding Flashcards | Quizlet**

Chapter 6 Chemical  
Bonds WordWise.

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## Chapter 6

Chemical Bonding

Section 2

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STUDY. Flashcards.

Learn. Write. Spell.

Test. PLAY. Match.

Gravity. Created by.

JolieDA. Key Concepts:

Terms in this set (12)

ionic. A type of bond that holds cations and anions together.

Covalent. A type of bond in which two atoms share a pair of valence electrons.

Molecule.

## **Chapter 6 Chemical Bonds WordWise**

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Chapter 6

Chemical Bonding  
**Flashcards | Quizlet**

IONIC BONDING

COVALENT BONDING

Atoms A Atoms B Atom

C Atom D Electrons

transferred from atoms

A to atoms B Electron

pair shared between

atom C and atom D +

+ Many atoms Two

atoms Atom C Atom D

Cation A Anion B + + +

+ + + +----- - SECTION

6.1 Nature favors

arrangements in which

potential energy is

minimized. For



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Chapter 6

Chemical Bonding

Section 1

Covalent Answer

Key

example, a boulder is less likely to balance

## **CHAPTER 6 Chemical Bonding**

Chapter 6: Chemical Bonding Section 1-

Introduction to

Chemical Bonding

Objectives: define

chemical bond;

differentiate between

covalent and ionic

bonding; explain why

bonding occurs; use

the difference in

electronegativity to

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Section 3  
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determine whether a bond is polar covalent, nonpolar covalent or ionic

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CHAPTER 6 REVIEW

Chemical Bonding

SECTION 3 SHORT

ANSWER Answer the following questions in the space provided. 1. a The notation for sodium chloride, NaCl, stands for one (a) formula unit. (c)

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## Chapter 6

Chemical Bonding

Section 2. d In a

crystal of an ionic  
compound, each cation

is surrounded by a  
number of (a)

molecules. (c) dipoles.

(b) positive ions. (d)

negative ions.

### **6 Chemical Bonding - Somerset Canyons**

#### 6.2 Covalent Bonding

The attractions

between the shared

electrons and the

protons in each

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## Chapter 6

Chemical Bonding

nucleus hold the atoms together in a covalent bond. • A covalent bond is a chemical bond in which two atoms share a pair of valence electrons. • When two atoms share one pair of electrons, the bond is called a single bond.

### **Chapter 6 Chemical Bonds**

Chapter 6 Chemical Bonding Section 4

Worksheet Answers

Chapter 6 Chemical

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Bonding Section Right  
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6 Chemical Bonding  
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Chemical Bonding**

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Chapter 6

Chemical Bonding

**Section 4 Worksheet**

**Answers**

Chapter 6 Chemical  
Bonds Section 6.2

Covalent Bonding

(pages 165–169) This section discusses the formation of covalent bonds and the factors that determine whether a molecule is polar or nonpolar.

**Chapter 6 Chemical  
Bonds Section 6.2  
Covalent Bonding**

Chapter 6 METALLIC

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## Chapter 6

Chemical Bonding

BONDING Section

4/11/19/2010

S.Martinez 49 50.

METALLIC BONDING

Explains why they This is due to the are such excellent highly mobile conductors of heat & valence electrons of electricity in the the atoms that make solid state compared up a metal.

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chapter 6 - LinkedIn  
SlideShare**

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## Chapter 6

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Bonding DRAFT. 8th -  
10th grade. 545 times.  
Chemistry. 76%

average accuracy. 3  
years ago. ayoung04.  
0. Save. Edit. Edit.

Chapter 6 Chemical  
Bonding DRAFT. ... Q.

Ionic bonds can be  
formed between two  
nonmetals. answer  
choices . True. False.

Tags: Question 10 .  
SURVEY . 30 seconds .

Q. What is the ionic  
compound formed



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Section 2

between ...

**Chapter 6 Chemical  
Bonding Quiz -  
Quizizz**

Holt McDougal Modern  
Chemistry Chapter 6:  
Chemical Bonding  
Chapter Exam  
Instructions. Choose  
your answers to the  
questions and click  
'Next' to see the next  
set of questions.

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Chemical Bonding

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