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Kscott_23. Terms in this set (46)

Chemical Reaction. A _____ is the
process by which one or more

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substances are changed into one or more different substances.

Chapter 8: Chemical Equations and Reactions Study Guide ...

Chapter 8: Chemical Equations 81 The Chemical Equation • A chemical equation represents a chemical reaction A balanced equation: o Summarizes the

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reaction, o Displays the substances that are reacting, o Shows the products, and o Indicates the amounts of all substances in the reaction • Let's

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CHAPTER 8 REVIEW Chemical Equations and Reactions SECTION 2 SHORT

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ANSWER Answer the following questions in the space provided. 1. Match the equation type on the left to its representation on the right. c synthesis (a) $AX + BY \rightarrow AY + BX$ d decomposition (b) $A + BX \rightarrow AX + B$ b single-displacement (c) $A + B \rightarrow AX$ a double-displacement (d) $AX \rightarrow A + X$

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Chemical Equations and Reactions
Flashcards ... $2 \text{Al} + 3\text{Fe}(\text{NO}_3)_2 \rightarrow 2\text{Al}(\text{NO}_3)_3 + 3 \text{Fe}$. the replacement of a metal in a compound by a more reactive metal, single replacement reaction.
identify the following reaction. $\text{Mg} + 2\text{HCl} \rightarrow \text{MgCl}_2 +$

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molecules on each side of a chemical
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Balancing Chemical Equations (Chapter
8) Chapter 8 Chemical Reactions
Worksheet 1. the equation must
represent known facts 2. the equation

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must contain the correct formulas for the reactants and products 3. the law of conservation of mass must be satisfied.

Chapter 8 Chemical Equations And Reactions Test Answer Key ...

Chapter 8 Chemical Reactions

Worksheet 1. the equation must represent known facts 2. the equation

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must contain the correct formulas for the reactants and products 3. the law of conservation of mass must be satisfied. A small whole number that appears in front of a 3. the law of conservation of mass must be satisfied.

Chapter 8 Lesson 2 Chemical Equations Reactions Answers ...

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Physical indicators of chemical reactions
Chapter 8 chemical equations and
reactions section 1 quiz answers. 1.
evolution of energy as heat and light(has
to be both), many physical changes also
involve either heat or Guidelines for
balancing chemical equations. Balance
the different types of atoms one at a
time.

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evolution of gas bubbles when two substances are mixed is often evidence of a chemical reaction. For example ...

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8-2 Balancing Chemical Equations Atoms rearrange in chemical reactions, but atoms cannot be created or destroyed (Law of Conservation of Matter). For this reason, the number of each kind of atom on the product and the reactant side must be the same.

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CHAPTER 10: CHEMICAL CALCULATIONS
and CHEMICAL EQUATIONS Part II ALEKS
Homework Chapter 10 Part II (8 Topics)
10.3 MOLARITY AND EQUATION
STOICHIOMETRY solution: homogeneous
mixture of substances present as atoms,
ions, and/or molecules solute:

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component present in smaller amount
solvent: component present in greater
amount SOLUTION CONCENTRATIONS A
concentrated solution has a large
quantity of ...

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