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solution slowly until some $\text{Na}_2\text{SO}_4 \cdot 10\text{H}_2\text{O}$ crystallizes out. Calculate the g of $\text{Na}_2\text{SO}_4 \cdot 10\text{H}_2\text{O}$ recovered in the crystals per 100 g of initial solution, if the residual solution after the crystals are removed contains 28% Na_2SO_4 . Right answer but: -10 if answer is in g of Na_2SO_4 and not g of $\text{Na}_2\text{SO}_4 \cdot 10\text{H}_2\text{O}$

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