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Practice 13 Name: _____ Date: _____

**Skill Practice 13 - High School
Guided Inquiry Chemistry ...**

Skill Practice 43 Name: _____ Date: _____

Hour: _____ 1. Give the following reaction
and data, answer the following

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questions. 2 CLO 2 + 2 I-2 CLO 2-+ I 2
Experiment [CLO 2] [I-] rate of
disappearance of I-1 2 3 0.30 0.90 0.30
0.40 0.40 0.80 3.350×10^{-4} M/s $9.045 \times$
 10^{-3} M/s 5.600×10^{-3} M/s

Skill Practice Combined - Scribnotes

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Practice 11 Name: _____ Date: _____

Skill Practice 11 - S.W.H.S
CHEMISTRY

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Contents (continued)

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© 2004 by Jason Neil. All rights reserved. Skill Practice 19 Name: _____
Date: _____ Hour: _____ 1. Write the symbol and charges for the following ions:

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Skill Practice 19 - S.W.H.S CHEMISTRY

AP Chemistry Review 2014 long
version.pdf What students are saying As
a current student on this bumpy
collegiate pathway, I stumbled upon
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online help from tutors 24/7, and even share my old projects, papers, and lecture notes with other students.

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Write answers in scientific notation. a)

$$(7.430 \times 10^4)(3.0 \times 10^2) = 2.229 \times 10^7$$

$$b) 8.03 \times 10^6 + 4.0 \times 10^4 = 8.07 \times 10^6$$

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$$6 \text{ c) } (2.22 \times 10^{-12}) / (4.10 \times 10^{-33}) = 5.41 \times 10^{20}$$
$$\text{d) } (35,020)(321.0) = 1.12 \times 10^7$$

Skill Practice 1

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a) Given your answers to questions 4-6,
what happens to the rate of the reaction

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if you increase the temperature of the reactants? b) Try to think of two reasons WHY and write them down.

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6. Considering your answer to question 5, how many electrons can fit in the entire 5th energy level? 7. Write the notation for an electron spinning

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clockwise in a p sublevel in the 4 th energy level.

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ChemQuests (Main Lessons) Skill
Practice. Chemistry Inquiry #13:

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Structure of Atoms. Skill Practice #13.
Chemistry Inquiry #16: Electron's
Address. Skill Practice #16. Chemistry
Inquiry #57: Intro. to Reaction Rates.
Skill Practice #57. Chemistry Inquiry
#81: Oxidation/Reduction

Sample Guided Inquiry Chemistry Lessons

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Perform the following operations and give the answers in the correct number of significant figures. If the question is in scientific notation, then please use scientific notation in your answer.

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Skill Practice 27 -

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Skill Practice 30 Conversion Practice

Name: Date: Hour: I. Calculate the number of molecules in 210 grams of 'water. 7.02×10^2 molecules 2. If you have 6.25×10^2 molecules of sulfur tetrafluoride, how many grams do you have? 1120 g 3. Define what a mole is.

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Don't give a number, give a definition in words.

bedfordjfhsharpschool.net

Calculate the average atomic masses. Round all answers to two decimal places. Record your answer with the correct units. 1. Iodine is 80% 127.1, 17% 126.1, and 3% 128.1. Calculate the

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average atomic mass of iodine. 2.
Calculate the average atomic mass of
lithium, which occurs as two isotopes
that have the following

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/ GPPS Home

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