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Modern Chemistry 47 Chapter Test Name Class Date Chapter Test A, continued _____ 7. The melting points of ionic compounds are higher than the melting points of molecular compounds because a. ionic substances tend to vaporize at room temperature. b. ionic substances are brittle. c. attractive forces between ions are greater than the attractive ...

Assessment Chapter Test A

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Assessment Chapter Test A - Wag & Paws

Modern Chemistry 81 Chapter Test. Name Class Date | Chapter Test B, continued PART III Write the answers to the questions on the line to the left, and show your work in the space provided. Use the table below to answer the questions. Element Symbol Atomic mass (amu) Bromine Br 79.90 Carbon C 12.01 Chlorine Cl 35.45

Name Class Date Assessment) Chapter Test B

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This chapter and the chapters that follow provide you with the tools necessary to answer these questions and many others like them. 9.1 A Typical Problem 9.2 Relating Moles to Number of Particles 9.3 Molar Mass and Chemical Compounds 9.4 Relationships Between Masses of Elements and Compounds 9.5 Determination of Empirical and

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