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Chemical formula. indicates the relative number of atoms of each kind in a chemical compound. Reveals the number of atoms of each element contained in a single molecule ...

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...

Compound name Aluminum sulfide Aluminum sulfite Formula

PbCl₂ (NH₄)₂PO₄ Hydroiodic acid SECTION 7-1 REVIEW

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Class CHAPTER 7 HOMEWORK 7-1 (pp. 203-206) VOCABULARY
Write true or false for each statement. 1.

Modern Chemistry • CHAPTER 7

CHAPTER 7 REVIEW Chemical Formulas and Chemical
Compounds SECTION 1 SHORT ANSWER Answer the following
questions in the space provided. 1. c In a Stock system name
such as iron (III) sulfate, the Roman numeral tells us (a) how
many atoms of Fe are in one formula unit. (b) how many sulfate
ions can be attached to the iron atom.

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ions can be attached to the iron atom. (c) the charge on each Fe ion.

7 Chemical Formulas and Chemical Compounds

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Chromium is "4 and each oxygen is #2. 56 SECTION 7-2
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7 REVIEW Chemical Formulas and Chemical Compounds
SECTION 7-3 SHORT ANSWER Answer the following questions in
the space provided. 1.

Chemical Formulas and Chemical Compounds

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CHAPTER 7 REVIEW Chemical Formulas and Chemical Compounds SECTION 1 SHORT ANSWER Answer the following questions in the space provided. 1. c In a Stock system name such as iron (III) sulfate, the Roman numeral tells us (a) how many atoms of Fe are in one formula unit. (b) how many sulfate ions can be attached to the iron atom.

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CHAPTER 7 REVIEW Chemical Formulas and Chemical Compounds MIXED REVIEW SHORT ANSWER Answer the following questions in the space provided. 1. Write formulas for the following compounds: CuCO₃ a. copper(II) carbonate Na₂SO₃ b. sodium sulfite (NH₄)₃PO₄ c. ammonium phosphate SnS₂ d. tin(IV) sulfide HNO₂ e. nitrous acid 2.

7 Chemical Formulas and Chemical Compounds

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CHAPTER 7 REVIEW Chemical Formulas and Chemical Compounds SECTION 1 SHORT ANSWER Answer the following questions in the space provided. 1. c In a Stock system name such as iron (III) sulfate, the Roman numeral tells us (a) how many atoms of Fe are in one formula unit. (b) how many sulfate ions can be attached to the iron atom.

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Section 4: 7 Chemical Formulas and Chemical Compounds.
CHAPTER 7 REVIEW Chemical Formulas and Chemical Compounds SECTION 4 SHORT ANSWER Answer the following questions in the space provided. 1. Write empirical formulas to match the following molecular formulas: CH₃O₂ a.

Modern Chemistry Chapter 7 Review Answers Section 4

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The Arrangement of Electrons in Atoms chapter of this Holt McDougal Modern Chemistry Companion Course helps students learn the essential lessons associated with how electrons are arranged in atoms.

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